

## Article

# Production, Preparation and Characterization of Microalgae-Based Biopolymer as a Potential Bioactive Film

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**Abstract:** Six microalgae strains were screened according to their biomass productivity and polymer synthesis, showing biomass productivity between 0.14 and 0.68 g/(L·d) for a 21-day growth period. Extracellular biopolymers from the spent culture media of *Nostoc* sp. (No), *Synechocystis* sp. (Sy), and *Porphyridium purpureum* (Pp) was obtained, and the yields of the clean biopolymer were 323, 204, and 83 mg/L, respectively. The crude biopolymer was cleaned up using a solid-phase extraction technique. The emulsification index  $E_{24}$  values for the clean biopolymer were 77.5%, 68.8%, and 73.3% at 0.323, 0.083, and 0.204 mg/mL, respectively. The clean biopolymer of the No strain showed the highest fungal growth inhibition against *Fusarium verticillioides* (70.2%) and *Fusarium* sp. (61.4%) at 2.24 mg/mL. In general, transparent and flexible biofilms were prepared using biopolymers of No and Pp. The microstructural analysis revealed the presence of pores and cracks in the biofilms, and the average roughness  $R_a$  values are 68.6 and 86.4 nm for No and Pp, respectively, and the root mean square roughness  $R_q$  values are 86.2 and 107.2 nm for No and Pp, respectively.

**Keywords:** microalgae; biopolymer; emulsification; antifungal; biofilms

## 1. Introduction

Nowadays, most packaging materials are still derived from nonrenewable sources. These synthetic polymers have been used as packaging materials for many years due to their economic and technological advantages, such as high availability, low cost, and favorable functional properties. However, such polymers present hydrophobic nature, which limits the action of microorganisms, and therefore takes many years to decompose. This results in the production of large volumes of solid wastes, which leads to serious environmental problems. These issues have been greatly aggravated due to the increase in population and economic growth from developed and developing countries. Hence, the need to reduce the amount of discarded plastics is being recognized globally, aiming to replace them with packaging films based on biodegradable materials, which are recognized as environmentally friendly materials [1].

Food, as the largest packaging user, and packaging industries have been joining efforts to reduce amounts of nonbiodegradable food packaging materials, since environmental issues are the major

concern of consumers. Therefore, several studies are being carried out, involving the development of new materials to produce packaging for short-term biodegradation. It is likely that the future generation of packaging materials will be derived from renewable and biodegradable resources [2].

Packaging plays a very important role in food preservation and in health-enhancing foods [3]. Food packages usually act as inert barriers for product protection with no interaction with food. However, edible films can provide additional protection for food, while being a fully biodegradable and environmentally friendly packaging system. Biofilms, as the primary barrier against physical impacts, prevent contamination, increase shelf life and contain important information about packaged food. In addition, biofilms can be used to improve food quality, as they can carry functional ingredients such as fatty acids, antioxidants, antimicrobials, nutrients, and flavors to further enhance food quality, stability, functionality, and safety [4].

The most common materials used in the formulation of edible films are proteins (e.g., gelatin, casein, wheat gluten, and zein) and polysaccharides (e.g., alginate, starch, and chitosan), which are used alone or blended. These biopolymers are highly biodegradable and decompose easily into inorganic by-products like carbon dioxide and water [5]. The use of vegetable raw materials to produce biodegradable packaging is then a favorable alternative.

Some species of microalgae (e.g., *Spirulina* sp., *Isochrysis galbana*, *Nannochloropsis oculata*, *Synechocystis* sp. (Sy), and *Nostoc* sp. (No)) can produce functional compounds with high added value, including essential fatty acids, pigments, proteins, polysaccharides, vitamins, and minerals. Due to the presence of biologically active compounds with antioxidant, antitumoral, anti-inflammatory, neuroprotective, and coloring properties in its biomass, microalgae are already being marketed as food supplements/additives (e.g., omega 3 and omega 6 fatty acids and pigments). In addition, microalgae are also being used to increase the nutritional content of conventional foods [6–10]. On the other hand, microalgae-based polymers have not yet been tested regarding the production of bioactive films and emulsification properties. Thereby, in this work, six microalgae strains (*Porphyridium purpureum* (Pp), No, *Spirulina maxima* (Sm), Sy, *Scenedesmus obliquus* (So), and *Spirogyra* sp. (Sp)) were screened to evaluate their potential for biopolymer production. Three of the biopolymers were selected, due to their extracellular biopolymer production; the antifungal and antibacterial activities as well as the emulsifying capacities and the antioxidant effects of the biopolymers were also evaluated.

## 2. Materials and Methods

### 2.1. Chemicals

The reagents used in this study were all of analytical grade, i.e., Albumin fraction V (Merck, Darmstadt, Germany), D-glucose anhydrous (Pronalab, Lisbon, Portugal), Trolox (97% 6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid) and ABTS (2, 2'-azino-bis-(3-ethylbenzothiazoline 6-sulfonic acid) diammonium salt) (Sigma-Aldrich, Toluca, Mexico).

### 2.2. Biological Materials

The strains of Pp SAG 1380-1a, No, Sm, Sy PCC 6803, So, and Sp SAG 170.80 were provided by the Laboratório Nacional de Energia e Geologia, Lisbon, Portugal. *Escherichia coli* ATCC 43895, *Xanthomona campestris* ATCC 13951, and the phytopathogenic fungi—*Botrytis cinerea* NRRL 1650, *Colletotrichum gloeosporioides*, *Curvularia* sp., *Fusarium* sp., *Fusarium verticillioides*, and *Helminthosporium* sp.—were obtained from Escuela Nacional de Ciencias Biológicas, Instituto Politécnico Nacional, Mexico City, Mexico.

### 2.3. Microalgae Culture

Pp, No, Sm, Sy, So, and Sp were cultivated in an autotrophic culture medium [11–13], using airlift bioreactors (capacity: 1L) with bubbling air at 1 vvm, a constant temperature of  $25 \pm 1$  °C,

and continuous white light illumination ( $100 \mu\text{E}/(\text{m}^2 \cdot \text{s})$ ) from fluorescence lamps (Philips TL-D 18W/54-765, Lisbon, Portugal).

A 10 mL sample of the culture was taken each 3 days to evaluate the growth rate by the optical density at 540 nm using a Hitachi U-2000 spectrophotometer. The pH of the medium was measured using a pH meter (INOLAB, WTW, Weilheim, Germany). The dry biomass productivity was evaluated every 7 days. Culture samples of 3.0–10.0 mL were taken and filtered through a GF/C filter with a  $1.2 \mu\text{m}$  pore size and dried at a constant temperature of  $80^\circ\text{C}$ , and the dry biomass concentration was calculated. The biomass productivity was expressed as dry biomass in grams per liter per day ( $\text{g}/(\text{L} \cdot \text{d})$ ).

After 21 days (at stationary phase), the agitation of the culture was stopped, and the culture was later centrifuged at 12,000 rpm for 10 min in order to harvest the biomass and separate the crude extracellular medium.

#### 2.4. Polymer Precipitation Capacity

Typically, microalgae-based polymers are extracted from biomass, and the culture medium was discarded. However, in this research, the presence of polymer was tested by selective precipitation from both the microalgae biomass extract and the crude extracellular medium. In the first scenario, one gram of the fresh microalgae biomass was suspended in 100 mL of 0.05% NaCl and heated at  $80^\circ\text{C}$  for 10 min with constant magnetic stirring. Then, the extract was filtered to vacuum with a Whatman filter,  $11 \mu\text{m}$  pore size. In the second scenario, the crude extracellular medium did not have additional treatment. The microalgae biomass extract and the crude extracellular medium were analyzed in terms of sugar and protein contents to obtain the biopolymer by precipitation.

For the precipitation test, a 10 mL sample of the microalgae biomass extract and the crude extracellular medium was mixed with 10 mL of a 0.5% aqueous cetrimide solution. The precipitate obtained was dialyzed (dialysis tubing cellulose membrane, typical molecular weight cut-off = 14,000) against 1 L of distilled water for a two-day period with daily water bath change. After dialysis, the precipitate obtained was used for biofilm preparation.

#### 2.5. Film-Forming Capacity

From the previous step, the crude extracellular medium and extract, which generated a precipitate, were used to produce biofilms using a sample of 0.3 g of the polymer precipitate. Due to scarcely available information concerning biofilm production from microalgae, five combinations were tested considering two solvents, two plasticizers, four preparation ways, and two drying strategies, resulting in the following formulations:

- The fresh polymer precipitate was dissolved in 10 mL of 5% NaCl with gentle agitation on a hot plate, until the temperature reached  $60^\circ\text{C}$ . The plasticizer (1.0 mL of glycerol) was added, and the agitation continued for two more minutes. Then, the viscous solution was precipitated with 0.5% cetrimide in a ratio of 1:1 ( $v/v$ ). The precipitate was recovered, laminated and dried at  $25^\circ\text{C}$  for 24 h.
- The fresh polymer precipitate was dissolved in 10 mL of 5% NaCl with gentle agitation on a hot plate, until the temperature reached  $60^\circ\text{C}$ . The plasticizer (0.5 g of D-sorbitol) was added, and the agitation continued for two more minutes. Then, the viscous solution was precipitated with 0.5% cetrimide in a ratio of 1:1 ( $v/v$ ). The precipitated was recuperated, laminated and dried at  $25^\circ\text{C}$  for 24 h.
- The fresh polymer precipitate was dissolved in 10 mL of 5% NaCl with gentle agitation on a hot plate, until the temperature reached  $60^\circ\text{C}$ . The plasticizer (1.0 mL of glycerol) was added, and the agitation continued for two more minutes. The viscous solution obtained was poured in a petri dish and dried at  $30^\circ\text{C}$  for 12 h.
- The fresh polymer precipitate was dissolved in 20 mL of 0.1 M NaOH with gentle agitation on a hot plate, until the temperature reached  $80^\circ\text{C}$ . The plasticizer (1.0 mL of glycerol) was added,

and the agitation continued for 20 min. The viscous solution was emptied in a petri dish and dried at 30 °C for 12 h.

- The fresh polymer precipitate was dissolved in 20 mL of 0.1 M NaOH with gentle agitation on a hot plate, until the temperature reached 80 °C. The plasticizer (1.0 mL of glycerol) was added, and the agitation continued for 20 min. The viscous solution obtained was precipitated with 0.5% cetrimide in a ratio of 1:1 (*v/v*). The precipitated was later recuperated, laminated and dried at 25 °C for 24 h.

## 2.6. Microstructure Analysis of the Biofilms

### 2.6.1. SEM

The surface microstructure was analyzed by SEM, and five surface images of the biofilms were obtained using Quanta™ 3D 250 FEG Scanning Electron Microscope (Thermo Fisher Scientific Inc., Eugene, OR, USA) at 30 kV and an environmental resolution.

### 2.6.2. Atomic Force Microscopy (AFM)

Samples of squares films (area: 1.0 cm<sup>2</sup>) were analyzed in a Veeco Atomic Force Microscope (Multimode V NanoScope, New York, NY, USA) connected to a controller (diNanoScope V, New York, NY, USA). Five areas of the film's surface (15 µm × 15 µm) were scanned using probes (RTEST) in the tapping mode at a scan rate of 1 Hz [14]. The roughness parameters, average roughness (Ra), and root mean square roughness (Rq) were obtained using NanoScope Analysis v7.30 software (Veeco, New York, NY, USA). The 3D images were observed in NanoScope Analysis v1.40 software (Veeco, New York, NY, USA).

## 2.7. Bioactivity of Microalgae Polymers

### 2.7.1. Biopolymer Clean-up

A clean-up process for the crude extracellular medium by solid phase extraction using C18 cartridges was implemented to obtain a clean biopolymer. In this step, hydrophobic compounds produced during algal growth cycles were eliminated.

Fifty milliliters of samples from the dialyzed extracellular medium were passed through a previously conditioned C18 column (ISOLUTE™ C18, 500 mg, 6 mL, Biotage, Charlotte, NC, USA). The nonretained fraction in the column, which corresponded to the clean extracellular medium, was collected and roto-evaporated to dryness at 40 °C. The yield of the cleaned and dried biopolymer was expressed as mg per liter of extracellular medium.

### 2.7.2. Antifungal Activity

All fungi strains used were activated on natural potato dextrose agar (PDA) prepared according to Sun et al. [15], using 0.37 mg/mL lincomycin and 0.37 mg/mL chloramphenicol to minimize contamination with bacteria. For inhibition assays, a 0.5 cm cylindrical portion of agar was removed from the center of the plate to create a cavity. An aliquot of 200 µL clean biopolymer with a concentration of 2.24 mg/mL dissolved in 3% NaCl was placed into the cavity. The plates were incubated at 28 °C for 24 h to allow for the diffusion of samples in the agar. Subsequently, the plates were inoculated with a 0.5 cm mycelium plug and incubated at 28 °C for 5 days. A negative control (3% NaCl) and a positive control (2.0 mg/mL fluconazole) were carried out simultaneously. All trials were performed in duplicate. The radial growth was measured, and the inhibition percentage was calculated using Equation (1):

$$\text{Inhibition (\%)} = \frac{\text{negative control area} - \text{treatment area}}{\text{negative control area}} \times 100\%, \quad (1)$$

$$\text{area} = \pi \cdot r^2. \quad (2)$$

### 2.7.3. Antibacterial Activity

*Escherichia coli* was activated in a liquid medium, Muller Hilton Broth, and then incubated at 35 °C for 24 h. *Xanthomonas campestris* was grown in plates using the yeast extract dextrose calcium carbonate agar and incubated at 28 °C for 24 h. The *Escherichia coli* inoculum was prepared according to Sadat et al. [16] through a dilution with a 0.85% NaCl sterile solution, until it reached turbidity equal to a 0.5 McFarland standard, which approximately corresponds to  $10^8$  CFU/mL. The *Xanthomonas campestris* inoculum was prepared through a dilution with a 0.85% NaCl sterile solution, until it reached an absorbance of 0.5 at 600 nm, which approximately corresponds to  $10^8$  CFU/mL [17]. Antibacterial activity was evaluated using the agar diffusion technique, where a 5  $\mu$ L inoculum was placed into the agar and expanded with a bacteriological loop. Then, a 0.7 cm sterile metallic cylinder was introduced into the agar, and a 25  $\mu$ L aliquot of the 2.24 mg/mL clean biopolymer dissolved in 3% NaCl was tipped, before the cylinder was quickly removed; aliquots of 25  $\mu$ L of 3% NaCl and 2.0 mg/mL lincomycin were used as negative and positive controls, respectively. The plates, in duplicate, were incubated at 28 °C for 24 h. The inhibition zone was measured and reported in centimeters.

### 2.7.4. Antioxidant Activity

The antioxidant activity of the clean biopolymer was determined by antiradical capacity using the ABTS radical technique [18]. Trolox dissolved in ethanol was used for the calibration curve. Seven milliliters of aqueous ABTS was mixed with 2.45 mM potassium persulfate and incubated for 16 h in the dark to generate free radicals. ABTS radical was diluted with ethanol to achieve an absorbance of  $0.7 \pm 0.05$  at 734 nm. A 100  $\mu$ L aliquot of the sample was added to 1.9 mL of diluted ABTS radical, vortexed for 10 s and kept in the dark for 7 min. The absorbance was measured at 734 nm, using as 100  $\mu$ L blank 3% NaCl, and similarly reacted with 1.9 mL ABTS radical. The samples of 2 mg/mL clean biopolymer dissolved in 3% NaCl were tested in triplicate.

### 2.8. Emulsification Index Test of the Extracellular Medium

Both the crude and clean extracellular media were tested to determine their emulsification indices at 24 h ( $E_{24}$ ) using kerosene as a hydrocarbon.  $E_{24}$  was determined by mixing 2 mL of kerosene and 2 mL of the sample in a 15 mL Falcon screw-capped tube. The tube was stirred during two minutes using a vortex, and the mixture was allowed to stand for 24 h.  $E_{24}$  was calculated by dividing the height of the emulsion layer by the mixture total height and then multiplying by 100 [19]. The samples were tested in triplicate.

### 2.9. Protein and Sugar Analysis

Lowry's method was used for protein analysis [20]. Bovine albumin serum was used as a standard for the calibration curve, and the absorbance was measured at 750 nm. The phenol-sulfuric acid method was used for the sugar analysis [21]. D-glucose was used as a standard for the calibration curve, and the absorbance was measured at 485 nm. Samples of 1.0 mL of the microalgae biomass extract and 1.0 mL of the crude extracellular medium were analyzed in triplicate. The results were expressed as mg D-glucose/g fresh biomass and mg protein/g fresh biomass for microalgae biomass extract and were expressed as mg D-glucose/L and mg protein/L for the crude extracellular medium.

### 2.10. FT-IR Analysis of the Clean Biopolymers

Five milligrams of the clean biopolymer powder was mixed with 50 mg potassium bromide powder, the mixture was compressed into a pellet, and its IR spectrum at wavenumbers from 500 to  $4000\text{ cm}^{-1}$  was recorded using a Spectrum 2000 FTIR spectrometer (Perkin Elmer, Inc., Waltham, MA, USA).

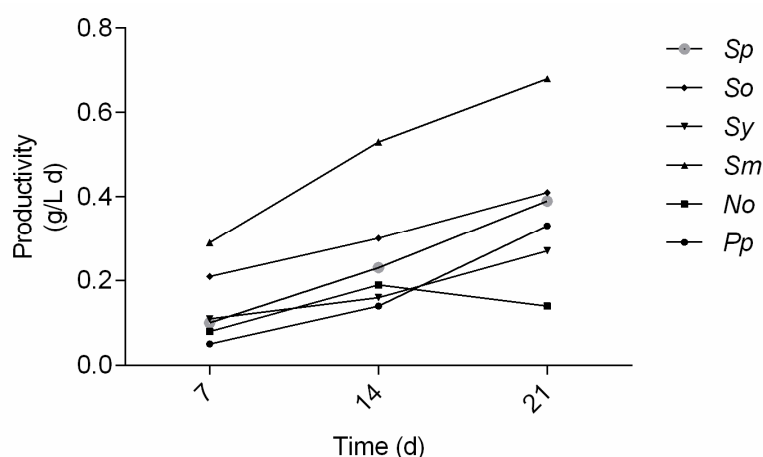
### 2.11. Statistical Analysis

Results from independent assays are reported as the mean  $\pm$  standard deviation. Means were compared by ANOVA and the Tukey's comparison test, with a significance level  $p < 0.05$ . Data were processed and analyzed in Minitab 17 (Minitab Inc., State College, PA, USA).

## 3. Results and Discussion

### 3.1. Microalgal Growth and Biomass Productivity

The strain with the fastest growth was Sm (Figure 1), with a maximum productivity of  $0.68 \text{ g}/(\text{L}\cdot\text{d})^{-1}$ , which was more than three times higher than the value reported by Barrocal et al. [22],  $0.20 \text{ g}/(\text{L}\cdot\text{d})^{-1}$ . The study was performed in autotrophic conditions for 21 days under light-dark cycles of 16 and 8 h separately. In addition, the productivity values obtained for the strains, So, Sp, and Pp, were similar ( $0.33\text{--}0.41 \text{ g}/(\text{L}\cdot\text{d})^{-1}$ ); however, the microalga No showed a decay in the last measurement. The use of microalgal biomass has been focused on obtaining high-value products as well as biofuels. Importantly, algae products and their high-value compounds (e.g., amino acids, carbohydrates, minerals, trace elements, phytohormones, sterols, and organic compounds and oils) are currently considered as alternatives to biofertilizers, biostimulants, [7,23], bioplastics [24–26], and liquid fossil fuels [27], among others.



**Figure 1.** Productivity ( $\text{g}/(\text{L}\cdot\text{d})^{-1}$ ). Pp: *Porphyridium purpureum*; No: *Nostoc* sp.; Sm: *Spirulina maxima*; Sy: *Synechocystis* sp.; So: *Scenedesmus obliquus*, Sp: *Spirogyra* sp.

### 3.2. Polymer Precipitation Capacity

Before the polymer precipitation test, sugar and protein contents were evaluated in the microalgae biomass extract and the crude extracellular medium. In general, the microalgae biomass extract presented sugar contents between  $2.6 \pm 0.0$  and  $10.3 \pm 0.1 \text{ mg/g}$  fresh biomass and protein contents between  $9.5 \pm 0.0$  and  $31.8 \pm 1.3 \text{ mg/g}$  fresh biomass, where So and No had the highest sugar and protein contents, respectively (Table 1).

**Table 1.** Sugar and protein contents in the microalgae biomass extract and the crude extracellular medium.

Microalgae	Microalgae Biomass Extract		Crude Extracellular Medium	
	Sugar <sup>(1)*</sup>	Protein <sup>(2)*</sup>	Sugar <sup>(3)*</sup>	Protein <sup>(4)*</sup>
Pp	9.6 ± 0.1 <sup>b</sup>	9.5 ± 0.0 <sup>c</sup>	0.09 ± 0.0 <sup>a</sup>	0.03 ± 0.0 <sup>d</sup>
No	8.6 ± 0.1 <sup>c</sup>	31.8 ± 1.3 <sup>a</sup>	0.03 ± 0.0 <sup>c</sup>	0.10 ± 0.0 <sup>c</sup>
Sm	2.6 ± 0.0 <sup>f</sup>	16.8 ± 0.2 <sup>b</sup>	0.02 ± 0.0 <sup>d</sup>	0.34 ± 0.0 <sup>a</sup>
Sy	7.4 ± 0.1 <sup>d</sup>	Not detected	0.04 ± 0.0 <sup>b</sup>	0.15 ± 0.0 <sup>b</sup>
So	10.3 ± 0.1 <sup>a</sup>	Not detected	0.02 ± 0.0 <sup>d</sup>	Not detected
Sp	5.0 ± 0.0 <sup>e</sup>	Not detected	0.05 ± 0.0 <sup>b</sup>	Not detected

Pp: *Porphyridium purpureum*; No: *Nostoc* sp.; Sm: *Spirulina maxima*; Sy: *Synechocystis* sp.; So: *Scenedesmus obliquus*; Sp: *Spirogyra* sp. <sup>(1)</sup> mg D-glucose/g fresh biomass; <sup>(2)</sup> mg protein/g fresh biomass; <sup>(3)</sup> mg D-glucose/L; <sup>(4)</sup> mg protein/L. Data are shown as mean ± SD (n = 3). \* Means with different letters are significantly different according to ANOVA and the Tukey test (comparison between strains) ( $p < 0.05$ ).

For the crude extracellular medium, the sugar content was in the range from  $0.02 \pm 0.0$  to  $0.09 \pm 0.0$  mg/L of the medium, and the protein content was from  $0.03 \pm 0.0$  to  $0.34 \pm 0.0$  mg/L of the medium. Pp showed the highest extracellular sugar content, and Sm presented the highest extracellular protein content (Table 1). These results are relevant, due to the limited information related to the exploitation of extracellular biopolymers from microalgae.

Although the microalgae biomass extracts contain sugar and protein, none of them were positive to the precipitation test. In contrast, the polymers precipitations from the crude extracellular media of Pp, No, and Sy were successful. Henceforth, these three strains in the culture medium were selected for the next studies.

According to Gloaguen et al. [28], an extracellular polysaccharide of *Porphyridium* sp. as a complex was precipitated due to ionic interaction. In addition, Andhare et al. [29] found that the precipitation method with quaternary ammonium salts at a concentration of 2% showed the minimum structural abrasion and a high efficiency during the precipitation of an exopolysaccharide (succinoglycan) from *Rhizobium radiobacter*. None of them demonstrated the production of biofilms from the biopolymer.

### 3.3. Film-Forming Capacity

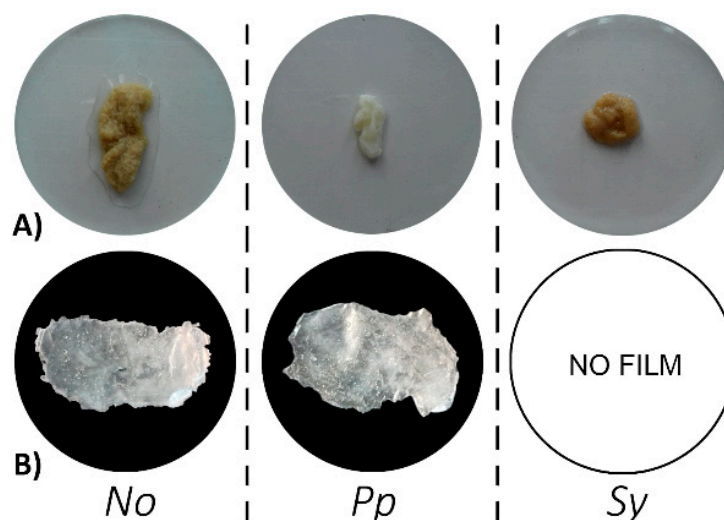
The precipitates from the crude extracellular media of Pp and No were able to form a film according to the proposed methods. Table 2 enlists the proposed formulations, as well as the preparation procedures. For instance, formulation I for both strains (Table 2) was selected as the best process conditions to obtain films (Figure 2).



**Table 2.** Characteristics of the biofilms of precipitates obtained from the crude extracellular medium.

Formulation	Solvent/ Plasticizer	Preparation/Drying	Biofilm Characteristics		
			No	Pp	Sy
I	5% NaCl/Glycerol	Agitation at 60 °C for 2 min precipitation and lamination at 25 °C for 24 h	Thin, transparent and flexible.	Thin, transparent and with mobility.	Does not form film
II	5% NaCl/D-sorbitol	Agitation at 60 °C for 2 min, precipitation and lamination at 25 °C for 24 h	Fragile	Thin, transparent and with mobility.	Does not form film
III	5% NaCl/Glycerol	Agitation at 60 °C for 2 min and emptying in petri dish at 30 °C for 12 h	Does not form film	Does not form film	Does not form film
IV	0.1 M NaOH/Glycerol	Agitation at 80 °C for 20 min and emptying in petri dish at 30 °C for 12 h	Does not form film	Does not form film	Does not form film
V	0.1 M NaOH/Glycerol	Agitation at 80 °C for 20 min and precipitation and lamination at 25 °C for 24 h	Thin, transparent, rough and flexible	Gross, rough and fragile	Does not form film

No: *Nostoc* sp.; Pp: *Porphyridium purpureum*; Sy: *Synechocystis* sp.



**Figure 2.** Precipitates (A) and biofilms from formulation I\* (B). No: *Nostoc* sp.; Pp: *Porphyridium purpureum*; Sy: *Synechocystis* sp. \* Conditions: fresh polymer precipitate dissolved in 5% NaCl, and were mixed with glycerol as a plasticizer and dried at 25 °C for 24 h.

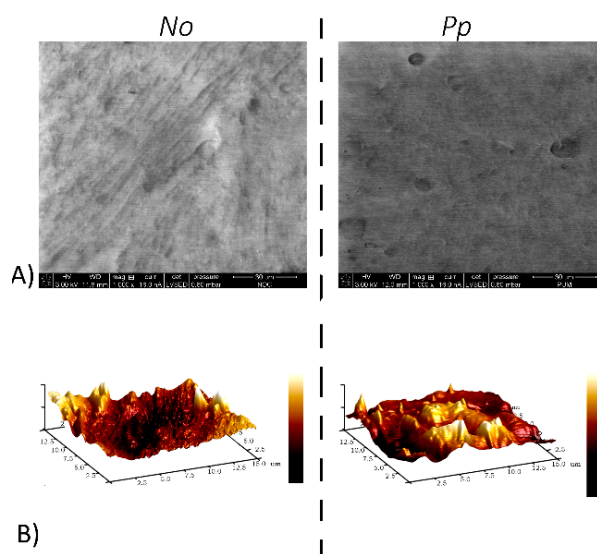
To the best of our knowledge, this is the first report about the isolation of polymers with the capacity to produce a transparent and flexible film from the exhausted microalgae in the culture medium. These qualitative characteristics are more promising in the field of packaging materials than those found by Benelhadj et al. [30], who reported the production of a fragile film with a dark brown color using a protein extracted from *Arthrospira (Spirulina) platensis* biomass. On the other hand, transparent biofilms are also useful in wound dressings due to the allowance of a better monitoring of the healing process [31].

The method proposed in this work involves the possibility to look for another quaternary ammonium salt in the film production, in order to obtain a similar potential application to that recently achieved by Demirci et al. [32]. Their research was focused on adding quaternary ammonium salts to a polymeric matrix of polyurethane, with the objective of allowing for the antimicrobial activity of the polymer, which is an application of wider utility.

By analyzing the morphological properties of the polymeric matrix, the SEM images of the surface (Figure 3A) showed cracks, fibrillary appearance, and presence of pores in the biofilm from No. The surface image of the biofilm from Pp was darker, and PP showed a smoother appearance



than No; nevertheless, it presented bigger and deeper pores. Bierhalz et al. [33] in their study with alginate–chitosan membranes (films) exhibited that the porosity allowed for the permeability of gases and water vapor, a favorable characteristic for its application in coverages of skin wounds through cell therapy. It is essential to mention that biopolymers (such as chitosan, sodium alginate, cellulose-based polymers, polylactic acid, and polyhydroxyalkanoates) are currently used as potential membrane materials in several technologies, such as microfiltration, ultrafiltration, pervaporation, gas separation, and membranes for tissue engineering [34,35]. In fact, the authors have mentioned the importance of getting nonanimal-origin biopolymers for biomaterial and tissue engineering fields [33]. The principal differences between the biofilm surfaces from No and Pp might be due to the type of network biopolymer formed, and although the laminate technique was the same, it could interfere in the structural conformation. Definitely, the film preparation procedure will play an important role in the physical properties of final polymer films [36].



**Figure 3.** SEM (A)\* and atomic force microscopy (B)\* images of the biofilms. No: *Nostoc* sp.; Pp: *Porphyridium purpureum*. \* indicates a selected image. n = 5.

The AFM topographic images (Figure 3B) of the biofilms showed irregular appearance, with agglomerates and high parameters roughness. No presented an average roughness value of  $R_a = 68.6$  nm and a root mean square roughness value of  $R_q = 86.2$  nm; for Pp, the values were shown as following:  $R_a = 86.4$  nm and  $R_q = 107.2$  nm. Higher roughness values imply greater contact surfaces, which could act as a carrier of antioxidants for active films [37] or be favorable to cell adhesion [31]. High roughness could affect film strength and barrier properties [38], and a film optimization is necessary for specific applications.

### 3.4. Clean Biopolymer

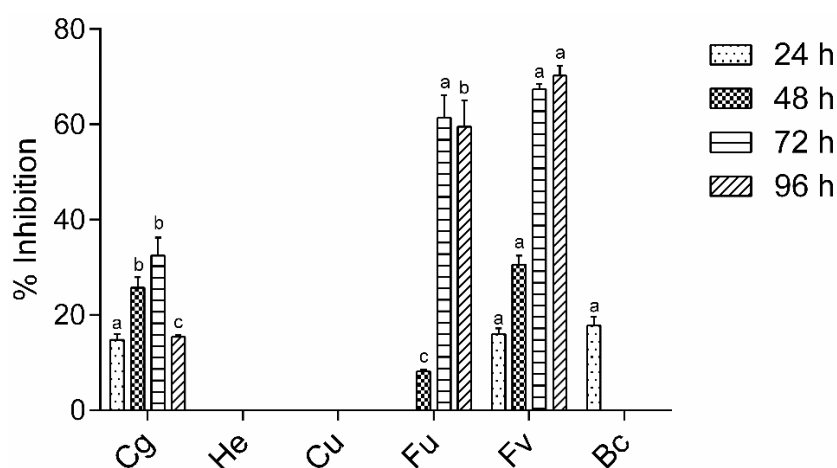
The yields of clean biopolymers of No, Sy and Pp was 323, 204 and 83 mg/L, respectively. The biopolymer production depends on the type of microorganism, culture conditions, type of bioreactor, and techniques of biopolymer isolation and purification [39]. Yeast, bacteria, and fungi are the main microorganism used for biopolymer production due to high yields obtained such as 6320–7960 mg/L from *Streptomyces carpaticus* [40], 461–737 mg/L from *Lactobacillus rhamnosus* [41], and 600 mg/L from *Aspergillus* sp. Y16 [42]. These yields are higher than those obtained with microalgae. Nevertheless, those microorganisms permanently require an organic carbon source for their growth. Microalgae, on the opposite, are capable of assimilating inorganic sources of carbon such as carbon dioxide, helping in regulating this greenhouse gas and transforming it into high-value compounds.

As mentioned above, the use of microalgae for producing high-value products and biopolymers is an environmentally friendly alternative; in order to increase the production yields of these products, it is necessary to continue with the selection for the best choice of strain and the optimization of culture conditions, such as light quality [43], light intensity, and nitrogen availability [44].

The FT-IR analysis was done using clean biopolymers. The infrared spectra of clean biopolymers from No, Pp, and Sy (Supplementary Figures S1–S3) showed absorption bands at 1260, 1255.5, and 1250  $\text{cm}^{-1}$ , respectively, attributed to stretching vibrations of sulfate ester groups. The absorption band was stronger for Pp than for No and Sy. Additionally, broadbands of O–H stretching at 3435.6, 3400.3, and 3412.3  $\text{cm}^{-1}$  were observed for No, Pp, and Sy respectively; an absorption band at 2930  $\text{cm}^{-1}$  was obtained for the three biopolymers due to stretching vibrations of C–H, and absorption bands at 1640.6, 1638.9, and 1655.7  $\text{cm}^{-1}$  were in the same order, corresponding to stretching vibrations of C=O from the carboxyl group, and bands at 1082.6, 1090.9 and, 1055.8  $\text{cm}^{-1}$  corresponded to stretching vibrations of asymmetric and symmetric C–O–C bonds and alcoholic C–O stretching vibrations, which indicated the presence of carboxylic acids [45] and carbohydrates [46].

### 3.5. Antifungal Activity

The antifungal activity test was done only with the No clean biopolymer, since this strain showed the best biopolymer yield after the clean-up process. Figure 4 shows the inhibition percentage against six fungal strains. The best growth inhibition was achieved against *Fusarium verticillioides*, a potential pathogen of maize [47], with the maximum inhibition rate of 70.2% at 96 h. *Helminthosporium* sp. and *Curvularia* sp. did not display growth inhibition by the biopolymer.



**Figure 4.** Antifungal inhibition (%) of NO's clean biopolymer (2.24 mg/mL) against *Colletotrichum gloeosporioides* (Cg), *Helminthosporium* sp. (He); *Curvularia* sp. (Cu); *Fusarium* sp. (Fu); *Fusarium verticillioides* (Fv), and *Botrytis cinerea* (Bc). Data are shown as mean  $\pm$  SE (n = 2). Means with different letters are significantly different according to ANOVA and the Tukey test ( $p < 0.05$ ). Comparisons between strains were performed for each time.

The studies done by Geresh et al. [48], concerning the biological activity of sulfated polysaccharides from marine microalgae *Porphyridium* sp. and *Rhodella reticulata*, revealed an inhibitory activity against neoplastic mammalian cell lines, and it was attributed to the negative charge of native sulfated and oversulfated exopolysaccharides and to the presence of glucuronic acid. Nevertheless, there are no data about the antifungal activities of the extracellular biopolymer microalgae of the strains reported in this study against the six phytopathogenic fungi. The present study shows that the No extracellular biopolymer could be an environmentally friendly alternative for fungi control amongst other already researched applications.

### 3.6. Antibacterial Activity

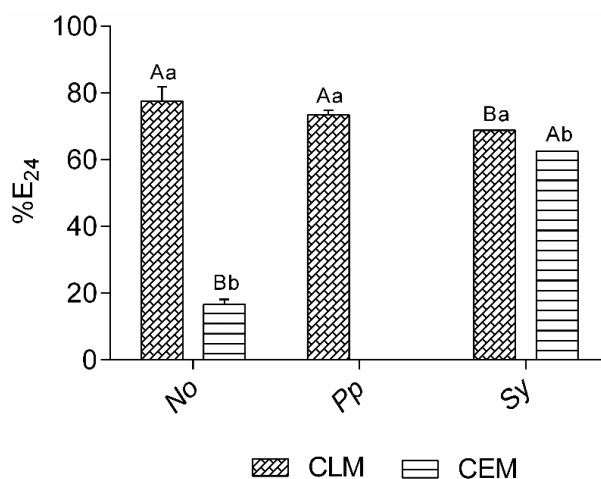
Neither *Escherichia coli* or *Xanthomonas campestris* showed growth inhibition for the two concentrations of the No clean biopolymer used. The results obtained in the present study for *Escherichia coli* were similar to that discovered by Najdenski et al. [49], where they did not find an inhibitory effect in the exopolysaccharides of *Nostoc entophytum* and *Nostoc muscorum*.

### 3.7. Antioxidant Activity

The clean biopolymers from the microalgae No, Pp, and Sy did not present antioxidant capacity. This means that the biopolymer were not capable of removing the ABTS radical generated in the aqueous phase. Sun et al. [50] found that the extracellular biopolymer (polysaccharide) of *Porphyridium cruentum* showed a weak antioxidant activity using the DPPH method, which was due to its high molecular weight and insolubility. Nevertheless, they observed that the hermetical microwave radiation to the polymer coupled to column separation allowed obtaining low-molecular-weight fragments, which could scavenge the radical generated with higher intensity than that of the nondegraded biopolymer.

### 3.8. Emulsification Capacity

The emulsification index ( $E_{24}$ ) results showed important differences between the crude extracellular medium and the cleaned one. Pp did not present emulsifying capacity in the crude extracellular medium, but in the clean one an  $E_{24}$  value of 73.3% was obtained (Figure 5). In the same way, the  $E_{24}$  value of No increased by 60.8%, and that of Sy presented a small increase of 6.3%. These results indicated that salts and compounds were eliminated during the cleaning process, avoiding the exposure of the hydrophilic part of the biopolymer, which permitted it to form the emulsion. The maximum emulsifying capacity was 77.5%, 73.3%, and 68.8% for No, Pp, and Sy, respectively, with concentrations of the clean biopolymers of 0.323, 0.083, and 0.204 mg/mL, respectively. The percentages found in this work are higher than the value of 59.1% found using surfactin ( $C_{14}/Lue_7$ ) produced by *Bacillus subtilis* with a concentration of 1.0 mg/mL and evaluated with kerosene [51]. It is also a higher value than the one found in the *Arthrospira (Spirulina) platensis* protein isolate, which was approximately 65% at pH 10 with a concentration of 100 mg/mL and evaluated using sunflower oil [30].



**Figure 5.** Emulsification index  $E_{24}$  (%). CLM: clean extracellular medium. CEM: crude extracellular medium. No: *Nostoc* sp.; Pp: *Porphyridium purpureum*; Sy: *Synechocystis* sp. Data are shown as mean  $\pm$  SE ( $n = 3$ ). Means with different letters are significantly different according to ANOVA and the Tukey test ( $p < 0.05$ ). Capital letters are for comparisons between strains, and small letters are for comparisons between treatments.

The microalgae biopolymer, as a biosurfactant or bioemulsifier, could play an important role in enhanced oil recovery [52] and the bioremediation of water or soil contaminated with hydrocarbon [53] and heavy metals. Moreover, the resulting biopolymer could find potential applications such as a nanoparticles stabilizer, an antimicrobial agent, and an additive in processed food, cosmetics, and textile industries [54].

#### 4. Conclusions

The importance of using biopolymers and biofilms with functionality is crucial in the modern societies. From the six microalgae strain screening, the biopolymers from the spent culture media of Py, NO, and Sy were produced and isolated. All polymers had remarkable emulsification capacities, although they did not display antioxidant activity. Importantly, this study showed, for the first time, that the biopolymers of Pp and No had the capacity to form rough biofilms with pores and cracks. Herein, further research should be performed on preparing compelling biofilms.

It is worth highlighting the considerable antifungal activity of No' extracellular biopolymer against *Fusarium verticillioides*, *Fusarium* sp. and small activities against *Colletotrichum gloeosporioides* and *Botrytis cinerea*. The results found in this work indicated that the microalgae biopolymer could be potentially useful in areas, such as food, feed, and cosmetic packaging, and additives in processed food, fungi control, and membrane materials, resulting in great development of food preservation and health-enhancing foods.

**Supplementary Materials:** The following are available online at <http://www.mdpi.com/2079-6412/10/2/120/s1>, Figure S1: IR spectrum of clean biopolymer of No sp.; Figure S2: IR spectrum of the clean biopolymer of Pp; Figure S3: IR spectrum of the clean biopolymer of Sy sp.

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